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Paper Authors

Dharmasoth Rama Devia, Ganga Rao Battua, Keloth Basavaiah



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Plant mediated green synthesized Magnetite Nanoparticles (Fe₃O₄ NPs) for Antioxidant, antibacterial, Anticancer activities-A review

Dharmasoth Rama Devi^{a*}, Ganga Rao Battu^a, Keloth Basavaiah^b

^a A.U. College of Pharmaceutical Sciences, Andhra University, Visakhapatnam-530 003, A.P, India.

^b Department of Inorganic and Analytical Chemistry, Andhra University, Visakhapatnam-530 003, A.P, India.

AUTHOR INFORMATION

Corresponding Author

Dharmasoth Rama Devi- A.U. College of Pharmaceutical Sciences, Andhra University, Visakhapatnam-530 003, A.P, India; Phone: +919398912208; Email:ramajoy90@gmail.com

Authors

Ganga Rao Battu - A.U. College of Pharmaceutical Sciences, Andhra University, Visakhapatnam-530 003, A.P, India; Email:ganga.battu@gmail.com

Keloth Basavaiah- Department of Inorganic and Analytical Chemistry, Andhra University, Visakhapatnam-530 003, A.P, India; Email:klbasu@gmail.com

Abstract

The usage of various plant extracts for green synthesis of magnetite nanoparticles, these plant extracts gaining importance day today when compared to the physical and chemical methods of synthesis due to its various advantages such as low cost, biocompatible, biodegradable, non-toxic. They also act as both reducing and capping agents during the synthesis of nanoparticles and this association achieved various pharmaceutical, and other biomedical applications. this study investigates the Plant mediated green synthesized Magnetite Nanoparticles (Fe₃O₄ NPs) for Antioxidant, antibacterial, Anticancer activities.

Keywords: plant-mediated Magnetite Nanoparticles (Fe₃O₄ NPs), Antioxidant, Antibacterial, Anticancer activities.

Introduction

During the past decade Fe₃O₄ NPs have attracted much attention for technological applications due to their unique properties, such as being superparamagnetic (Mahdavian and Mirrahimi, 2010), biocompatible, biodegradable, and non-toxic to humans (Hu, F. Q *et al.*, 2006; Zhao, H *et al.*, 2009; Zhang, L *et al.*, 2013). These unique properties allow Fe₃O₄ NPs to be widely used in different areas of applications, such as catalysis (Gawande *et al.*, 2013; Sharad *et al.*, 2014) magnetic storage media, (Terris and Thomson, 2005), biosensors, magnetic resonance imaging (MRI) (Kavitha *et al.*, 2013; Haw *et al.*, 2010) and targeted drug

delivery (Qiao *et al.*, 2009; Salem *et al.*, 2015; Li, X *et al.*, 2012; Wani *et al.*, 2014).

Various methods have been

employed for the synthesis of Fe₃O₄ NPs such as sol-gel method (Lemine *et al.*, 2012), solid-state synthesis (Paiva *et al.*, 2015), and flame spray synthesis (Kumfer *et al.*, 2010). Green synthesized Fe₃O₄ NPs can possess better characteristics, such as higher biocompatibility and biodegradability compared to physically synthesized Fe₃O₄NPs. Hence, they can be utilized in biomedical applications due to the special surface coating of green materials, which is

not only non-toxic and biocompatible yet also allow targeted drug delivery with Fe₃O₄ NPs localization in a particular area. Toxicity towards the human body can be minimized because the green materials used for synthesizing Fe₃O₄ NPs are safe to be consumed and thus it would be beneficial in biomedical applications. In contrast to the time-consuming chemical and physical methods which involve complicated procedures, the green method is much easier and safer to use, and especially plant-mediated synthesis of Fe₃O₄ NPs is a new and green method. The phytochemicals present in plants are capable of synthesizing crystalline Fe₃O₄ NPs. (Yuvakkumar and Hong, 2014) synthesized Fe₃O₄ NPs with particle size 100-200 nm from Fe(NO₃)₂·6H₂O as precursor using peel waste extract of Rambutan as a green ligation and chelating agent. Valentin *et al.*, (Makarov *et al.*, 2014) have prepared spherical shaped Fe₃O₄ NPs by using FeCl₃·6H₂O as a precursor and aqueous extracts of *Hordeum vulgare* and *Rumex acetosa* plants as reducing and capping agents. Sathishkumar *et al.*, have synthesized inverse cubic spinel structured, polydispersed and spherical shaped Fe₃O₄ NPs via co-precipitation method by using FeCl₃·6H₂O as a precursor and an aqueous fruit extract of *C. guianensis* as reducing and capping agents for antibacterial activity and cytotoxic activity (Sathishkumar *et al.*, 2018). Prasad *et al.*, have synthesized Fe₃O₄ NPs from FeCl₃

6H₂O using extract of watermelon rinds as a solvent and capping and reducing agent for catalytic activity (Prasad *et al.*, 2016). Latha *et al.*, have prepared orthorhombic structured spherical shaped Fe₃O₄ NPs from FeCl₃·6H₂O using leaf extract of *Caricaya papaya* as

reducing and capping agents (Latha and Gowri, 2014). Patra *et al.*, have prepared Fe₃O₄ NPs via co-precipitation of FeCl₂ and FeCl₃ using an aqueous ear leaf *corn* extract for antioxidant and antibacterial activities (Patra *et al.*, 2017). Khataee *et al.*, (Khataee *et al.*, 2017) have prepared Fe₃O₄ NPs by co-precipitation of FeCl₃·6H₂O and FeCl₂·4H₂O in presence of coffee waste hydrochar extraction as a capping agent. Lunge *et al.*, have successfully synthesized cubic structured Fe₃O₄ NPs with the particle size ranges from 5-25 nm by precipitation of FeCl₃·6H₂O in presence of tea waste template as capping agent (Lunge *et al.*, 2014). Horst *et al.*, have prepared Fe₃O₄ NPs by co-precipitation of 2:1 molar ratio of FeCl₃·6H₂O and FeSO₄·7H₂O using Gum Arabic as capping agent used for hyperthermia treatments (Horst *et al.*, 2017). Niraimathee *et al.*, have prepared spherical shaped Fe₃O₄ NPs with an average particle size of 67 nm from FeSO₄ using an aqueous root extract of *Mimosa pudica* as a capping agent used for targeted drug delivery (Niraimathee *et al.*, 2016). Buazar *et al.*, have prepared Fe₃O₄ NPs with an average particle of 40 nm Fe₃O₄ NPs from FeSO₄·7H₂O as the precursor using homemade starch-rich potato aqueous extract as the reducing agent and a stabilizing layer for the catalytic activity towards the removal of a methylene blue organic dye contaminant in wastewater (Buazar *et al.*, 2016). Bahadur *et al.*, have reported a modified co-precipitation method for the synthesis of water-soluble citric acid modified ultrafine superparamagnetic ferrofluid with 12 nm Fe₃O₄ NPs from FeCl₃·6H₂O and FeCl₂·4H₂O salts solution using hydrazine (N₂H₄) and sodium hydroxide in presence of lemon juice (Bahadur *et al.*, 2017). Venkateswarlu *et al.*, successfully synthesized spherical shaped Fe₃O₄ NPs with

the particle size ranging from 9-20 nm from FeCl₃ 6H₂O as a precursor in presence of seed extract of *Syzygium cumini* (Venkateswarlu *et al.*, 2014). Kumar *et al.*, synthesized spherical shaped Fe₃O₄ NPs and with the average particle size of 54.5 ± 24.6 nm from FeSO₄ 7H₂O as the precursor using leaf extract of *Andean blackberry* (Kumar *et al.*, 2016). Phumying *et al.*, successfully synthesized crystalline and inverse cubic spinel structured Fe₃O₄ NPs with the particle size ranging from 6-30 nm by hydrothermal reduction of (Fe(C₅H₈O₂)₃) as the precursor in presence of *Aloe vera* plant extract solution for targeted drug delivery (Phumying *et al.*, 2013). Rajendran *et al.*, synthesized crystalline and

cubic structured Fe₃O₄ NPs with an average particle size from 25-60 nm from FeCl₂ as precursor using leaf extract of *Sesbania grandiflora* (Rajendran and Sengodan, 2017). Alavi, M., Karimi, N., & Valadbeigi, T. (2019) reported the antibacterial, antibiofilm, antiquorum sensing, antimotility, and antioxidant activities of green fabricated Ag, Cu, TiO₂, ZnO, and Fe₃O₄ NPs via *protoparmeliopsis muralis* lichen aqueous extract against multi-drug-resistant bacteria. Biswas *et al.* , reported *Mikania mikrantha* leaf extract mediated biogenic synthesis of magnetic iron oxide nanoparticles: Characterization and its antimicrobial activity study. *Materials* (Biswas *et al.* , 2021).

Table 2.3. Different sizes, shapes, and applications of synthesized Fe₃O₄ NPs by using plants

Plant name	Part used	Size (nm)	Shape	Application	References
<i>Leucas aspera</i>	Leaves	20	Irregular rhombic	antibacterial activity and antioxidant studies	Wani <i>et al.</i> , 2014
<i>C. guianensis</i>	Fruit	17	Spherical	Antimicrobial Cytotoxic activity	Paiva <i>et al.</i> , 2015
<i>Caricaya papaya</i>	Leaves	25	Spherical	Biomedical applications	Yuvakkumar and Hong, 2014
<i>Syzygium Cumini</i>	Seed	9–20	Spherical	bio-medicinal	Khataee <i>et al.</i> , 2017
<i>Andean blackberry</i>	Leaf	54.5	Spherical	Antioxidant activity	Lunge <i>et al.</i> , 2014
<i>Amaranthus spinosus</i>	Leaf	91-125	spherical shape with rhombohedral	photocatalytic and antioxidant activity	Muthukumar and Matheswaran, 2015

<i>Sageretia thea</i> (Osbeck)	Leaf	473	Rhombohedral	Antioxidant, antibacterial, Anticancer,	Khalil <i>et al.</i> , 2017
<i>Grewia optiva</i> and <i>Prunus persica</i> phyto species:	Leaf	15-60 and 13-70	Quasi-spherical and Spherical	antibacterial and antioxidant activity	Mirza <i>et al.</i> , 2018

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