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**Paper Authors** 

N. SRINIVASA RAJNEESH, K. BHARADWAJA





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# IMPACT OF FORMALDEHYDE ADDITION ON AUTO-IGNITION IN INTERNAL-COMBUSTION ENGINES

<sup>1</sup>N. SRINIVASA RAJNEESH, <sup>2</sup>K. BHARADWAJA

<sup>1,2</sup> Assoc.Professor, Department of Mechancial Engineering, Malla Reddy Engineering College (Autonomous), Maisammaguda(H), Gundlapochampally Village, Secunderabad, Telangana State -500100

By employing a direct-injection diesel engine equipped with a common-rail type of in-jection system, by adding formaldehyde (CH2O) to the intake air, and by changing the fuel-injection timing, the compression ratio and the intake-air temperature, a mechanism for CH2O as a fuel additive to affect auto-ignition was discussed. Unlike an HCCI type of engine, the diesel engine can expose an air-fuel mixture only to a limited range of the in-cylinder temper- ature before the ignition, and can separate low- and high-temperature parts of the mechanism. When low-temperature oxidation starts at a temperature above 900 K, there are cases that the CH2O advances the ignition timing. Below 900 K, to the contrary, it always retards the tim- ing. It is because, above 900 K, a part of the CH2O changes into CO together with H2O2 as an ignition promoter. Below 900 K, on the other hand, the CH2O itself acts as an OH radi- cal scavenger against cool-flame reaction, from the beginning of low-temperature oxidation. Then, the engine was modified for its extraordinary function as a gasoline-knocking gener- ator, in order that an effect of CH2O on knocking could be discussed. The CH2O retards the onset of auto-ignition of an end gas. Judging from a large degree of the retardation, the ignition is probably triggered below 900 K.

**Key Words**: Internal Combustion Engine, Ignition, Combustion, Auto-Ignition, Spray Combustion, Knocking, Ignition Control, Additive, Formaldehyde, Flame Light Emissions

#### 1.Introduction

There are auto-ignition dominated subjects of combustion in internal combustion engines, that is, those of combustion control for homogeneous-charge compression-ignition (HCCI) types of engines, knocking control for gasoline engines, and new combustion modes for diesel engines to reduce both NOx and smoke emis- sions. Especially of HCCI engines, a lot of researchers

have been making efforts to resolve one of its difficult problems, how to control the ignition timing and stabi- lize the sensitive timing, as they recognized its potential advantages of a high thermal efficiency and very-low NOx and zero smoke emission levels. Effects of directly-controllable factors such as fuel properties and engine-operating conditions of the intake-air temperature, the compression ratio, the airfuel ratio and the EGR ratio on the ignition



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have been investigated. As another approach to HCCI timing control, ideas of making use of a fuel ad- ditive, for example, hydrogen peroxide(1) or ozone(2) as an ignition promoter, and formaldehyde(3) – (5) or methanol(2) as an ignition retardant, have been proposed. By Aceves et al. with the aid of calculation involving one-dimensional hydrodynamics transport and detailed chemical kinetics, hundreds of potential additives were ranked according to their capabilities for advancing the ignition timing(6).

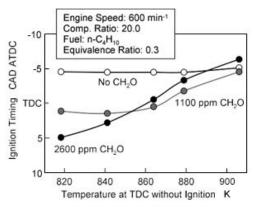


Fig. 1 E□ect of CH<sub>2</sub>O addition on ignition timing of air n-C<sub>4</sub>H<sub>10</sub> mixture<sup>(4)</sup>

#### 2.Experimental Procedure

2.1 Test engine and its operating conditions A four-valve single-cylinder naturally-aspirated direct-center-injection diesel engine equipped with a common-rail type of injection system, was supplied for this study. A pressure transducer (AVL GU12S-10) was mounted on the cylinder head, in place of the glow plug. The intake-air temperature was measured 70 mm upstream of an intake valve. A part of the exhaust gas was by-passed into an exhaust-gas analyzer (Horiba MEXA-8220D). Table 1 lists specifications of the engine together with its

operating conditions. In a conventional use of the engine fueled with gas oil (cetane index (JISK2280): 52), the compression ratio was set at 18.0 or

12.3. The fuel-injection timing was changed between 30 and 5 crank-angle degrees (CAD) ATDC. The injection timing of 30 CAD ATDC is close to the limit of the advancement, for fuel sprays out of the multi-hole nozzle to be captured within the piston cavity. But also, conve- niently, it is almost coincident with a timing before which any cool flame could not appear at the low compression ratio, even in case of HCCI. There was also a case at the low compression ratio, that the intake-air temperature was increased by 50 K using a heater installed upstream of the intake manifold, in order that the in-cylinder temperature at TDC only with the air could be almost the same as that at the high compression ratio without any heating. In these ways, a temperature to which an airfuel mixture was exposed before the autoignition, was changed between 660 and 930 K at a rough estimate with the intake-air temperature without any consideration of a residual gas and a heat loss. In another case at the low compression ratio, O2 was added to the intake air, and its level in the charge was increased up to 26.0%, so that the difference could be discussed, when the ignition delay was shortened by the high temperature or by the high O2 level.



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Table 1	Fnoine	specifications	and o	nerating	conditions

	CI Diesel C	SI Gasoline Combustion				
Bore / Stroke / Displacement	8					
Engine Speed						
Comp. Ratio (Geometrically)	12.3 (11.4)	18.0 (16.7)	11.0 (9.5)			
Combustion Chamber	Reentrant Type of Cavity (Min. Dia. / Max. Dia. / Depth: (Min. Dia. / Max. Dia. / Depth: 46.0 / 54.9 / 23.2 mm)		Pancake Type			
In-Cylinder Flow	Swirl-Control Valve Closed (Estimated Swirl Ratio of 3.7 at Differential Press. of 0.004 MPa through Steady-Flow Rig)					
Fuel	Gas Oil (Cetane Inc	Regular Gasoline (RON: 90.3				
Fuel Preparation		Intake-Port Injector				
Location of	Nozzle: on Cylinder Axis		Plug: 18 mm off Cylinder Axi			
Timing of	Injection: -30, -20, -10 and -5 CAD ATDC		Ignition: -10 CAD ATDC			
Air Charge	NA, WOT, No EGR	NA, WOT, 35% Charge, No EGR	65% Charge, No EGR			
Charge Temperature in Intake Port	305 K, 355 K (Heating)	305 K, 312 K (35% Charge)	308 K			
Coolant Temperature						
O <sub>2</sub> Level in Charge	Atmospheric, Rich (26 %)	Atmospheric	Atmospheric			
CH <sub>2</sub> O Level in Charge	1900 +/- 100 ppm	1900 +/- 100 ppm, 5000 ppm (35% Charge)	3000 ppm			
Air or O <sub>2</sub> Excess Ratio Excluding CH <sub>2</sub> O	3.4, 3.3 (Heating), 4.2 (26% O <sub>2</sub> )	3.4, 1.2 (35% Charge)	Stoichiometric			
Temperature at Injection Timing	610, 700, 770 and 800 K, 720, 810, 900 and 930 K (Heating)	660, 770, 880 and 930 K	-			

### 2. 2 Formaldehyde feed

CH2O-feeding conditions were set equally to those in the previous studies. In the studies, a part of in-cylinder gas had been directly sampled, and the level of CH2O in it had been measured using a gaschromatograph equipped with a flame ionization detector (GL Science GC-390TTDF)(3), (4). The previous data could be used for calibration on the level of CH2O in this study.

The amount of CH2O was monitored also by means of its value derived from the difference in the total amount of C atoms in exhaust emissions with or without the CH2O addition. Its validity is dependent on a previous result that, even in an extremelylean HCCI engine, almost all of added CH2O is oxidized into CO and CO2 during the combustion, which are detectable using the exhaust-gas analyzer, as shown in Fig. 2(4). Actually, however, in many cases at the low compression ratio the conventional use of the engine, the indicated total level of HC emissions with the CH2O was higher than that without any addition. The difference was less than 200 ppm, although the value is unreliable.

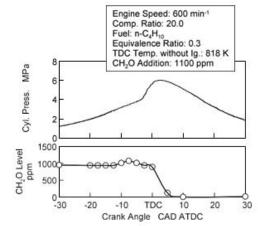


Fig. 2 History of in-cylinder CH<sub>2</sub>O level in case of HCCI engine fueled with n-C<sub>4</sub>H<sub>10</sub><sup>(4)</sup>

a momentary engine operation at the high O2 level. After all, there was no big difference in the value based on the C counting or the calibration with the previous data.

In some engine-operating conditions, the indicated total level of HC emissions with the CH2O addition was higher than that without any addition, although the value of the difference is unreliable. The CO level with the CH2O was always higher than that without any addition. A part of the CH2O, distributed in local areas off from combustion, seems to be emitted as being something ex- cept for CO2 and H2O.

#### 3. Results and Discussion

3. 1 Effect on formaldehyde addition on auto- ignition in diesel engine

Figures 3 to 5 show histories of the incylinder pres- sure and the effective heat release rate (EHRR) with or without the CH2O addition, at the different compression ratios and charge temperatures,



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ev- ery 0.125 CAD over continuous 30 cycles and cycle- averaged. Figure 6 shows ignition delays from the fuel- injection timings. Here, the ignition timing was defined as the timing when the gradient of an EHRR history first reaches 0.02 MPa/CAD2 (12.4 J/CAD2) in an empirical way. Cool-flame reaction or faint heat release before the ignition appears at the same timing with or without the CH2O, as drawn with a single dotted line. At the low com- pression ratio, the CH2O weakens the pre-ignition heat re- lease, and retards the ignition timing with any injection timings, compared to that without any addition. This is remarkable with the earliest injection timing when a two- stage type of heat release characterized by cool-flame reaction, appears.

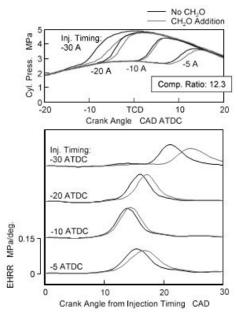


Fig. 3 Histories of in-cylinder pressure and e Dective heatrelease rate in case of low compression ratio

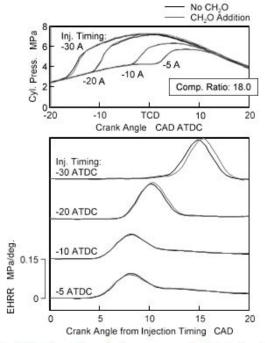


Fig. 4 Histories of in-cylinder pressure and e∏ective heatrelease rate in case of high compression ratio

3. 2 When ignition delay is deceived For an extended understanding of the opposite effects of the CH2O addition, contrastive cases to the above ones were discussed. One that the ignition delay was short- ened by the O2 addition to the intake air, even at the low compression ratio without any assistant rise in the charge temperature, to the extent of the delay shortened by the charge heating. The other that the delay was extended by the intake-air throttling, even at the high compression ratio, to the extent of that at the low compression ratio.

Figure 7 shows ignition delays as well as histories of the in-cylinder pressure and the EHRR, at the high O2 level in the charge.



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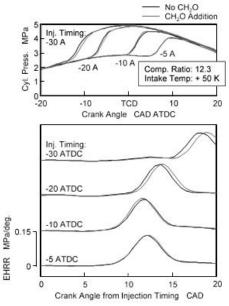
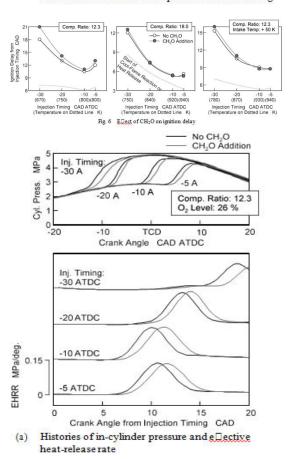


Fig. 5 Histories of in-cylinder pressure and except heatrelease rate in case of low compression ratio and heating



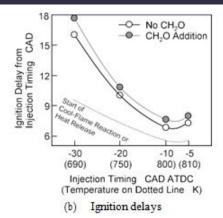


Fig. 7 Ellect of CH<sub>2</sub>O on ignition delay in case of low compression ratio and high O<sub>2</sub> level

#### 4. Conclusions

For close discussion about a mechanism for CH2O as a fuel additive to interact with auto-ignition of an air-fuel mixture, a directinjection diesel engine equipped with a common-rail type of injection system was employed. It is because, unlike an HCCI engine, it can expose a mix- ture only to a limited range of the in-cylinder temperature before the ignition, and therefore, can separate lowtemperature and temperature parts of interaction of CH2O, that is, parts interacting with cool-flame reaction and blue-flame reaction. Vapor of CH2O was added to the intake air. The ignition was induced at different fuelinjection timings, compression ratios and intake-air tem- peratures, that is, at different in-cylinder temperatures. Light emission spectra at the ignition timing were detected. When low-temperature oxidation starts at a tempera- ture above 900 K, there are cases that the CH2O addition advances the ignition timing. Below 900 K, to the con- trary, it always retards the timing. In case of the remark- able retardation, the



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CH2O completely eliminates the OH radical emission. After all, the experimental results of the ignition timing are consistent with those using an HCCI engine in Fig. 1. This means that in case of the advance- ment of the HCCI timing, the interaction of CH2O with cool-flame reaction is negligible for the timing.

The opposite effects of the CH2O on the ignition de- lay were most simply modeled. Above 900 K, a part of the CH2O is changed into CO and H2O2, and the latter acts as an OH radical generator in the final stage of low-temperature oxidation. Below 900 K, on the other hand, the CH2O itself acts as an OH radical scavenger against cool-flame reaction, from the beginning of low-temperature oxidation.

Then, the engine was modified for its extraordinary function as a gasoline-knocking generator, in order that an effect of CH2O on knocking could be discussed. The CH2O retards the onset of auto-ignition of an end gas. Judging from a large degree of the retardation, the igni- tion is probably triggered below 900 K.

#### References

- (1) Mulenga, M.C., Ting, D.S.-K., Reader, G.T. and Zheng, M., The Potential for Reducing CO and NOx Emissions from an HCCI Engine Using H2O2 Addi-tion, SAE Paper 2003-01-3204.
- (2) Yamada, H., Yoshii, M. and Tezaki, A., Chemical Mechanistic Analysis of Additive Effects in Homo-geneous Charge Compression Ignition of Dimethyl Ether, Proceedings of the Combustion Institute, Vol.30 (2005), pp.2773–2780.

- ( 3 ) Furutani, M., Isogai, T. and Ohta, Y., Ignition Charac- teristics of Gaseous Fuels and Their Difference Elimi- nation for SI and HCCI Gas Engines, SAE Paper 2003-01-1857. Subramanyam Pavuluri https://orcid.org/0000-0002-2703-2358
- (4) Yamaya, Y., Furutani, M. and Ohta, Y., Premixed Compression Ignition of Formaldehyde-Doped Lean Butane/Air Mixtures in a Wide Range of Temperature, SAE Paper 2004-01-1977.
- (5) Konno, M., Chen, Z. and Miki, K., Computational and Experimental Study on the Influence of Formaldehyde on HCCI Combustion Fueled with Dimethyl Ether, SAE Paper 2003-01-1826.
- ( 6 ) Aceves, S.M., Flowers, D., Martinez-Frias, J., Espinosa-Loza, F., Pitz, W.J. and Dibble, R., Fuel and Additive Characterization for HCCI Combustion, SAE Paper 2003-01-1814.
- (7) Kuwahara, K., Watanabe, T., Shudo, T. and Ando, H., A Study of Combustion Characteristics in a Direct In- jection Gasoline Engine by High-Speed Spectroscopic Measurement, The 13th Internal Combustion Engine Symposium, Japan, (in Japanese), (1996), pp.145–150.
- (8) Fischer, S.L., Dryer, F.L. and Curran, H.J., The Reac- tion Kinetics of Dimethyl Ether. I: High-Temperature Pyrolysis and Oxidation in Flow Reactors, International Journal of Chemical Kinetics, Vol.32 (2000), pp.713–740. https://orcid.org/0000-0002-2703-2358
- (9) Curran, H.J., Fischer, S.L. and Dryer, F.L., The Reac- tion Kinetics of Dimethyl Ether. II: Low-Temperature Pyrolysis and



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Oxidation in Flow Reactors, International Journal of Chemical Kinetics, Vol.32 (2000), pp.741–759.